Docket No.: 0171-0811P

AMENDMENTS TO THE CLAIMS

1. (Canceled).

2. (Currently amended) A method for preparing carbonaceous material powder mixture for electrical double-layer capacitors by placing an activated carbon and at least one electrically conductive powder selected from the group consisting of carbon black, Ketjen black, acetylene black, carbon whiskers, carbon fibers, natural graphite and synthetic graphite in a mixing container comprising a planetary mixer, wherein said method comprises a step consisting of rotating and revolving the container so as to form a dry mixture.

wherein the activated carbon has an average particle size of 1 to 100 μm, and wherein the conductive powder adheres to a periphery of the activated carbon and has an average particle size that is 10 nm to 10 μm and is smaller than the average particle size of the activated carbon.

3. (Previously presented) The method of claim 2, wherein the powder mixture is composed of 0.1 to 20 parts by weight of the conductive powder per 100 parts by weight of the activated carbon.

4-12, (Canceled).

13. (Withdrawn) A carbonaceous material powder mixture for electrical double-layer capacitors, which powder mixture comprises:

activated carbon with an average particle size of 0.1 to $100 \, \mu m$, and an electrically conductive powder which adheres to the periphery of the activated carbon:

wherein the conductive powder is at least one powder selected from the group consisting of earbon black, Ketjen black, acetylene black, carbon whiskers, carbon fibers, natural graphite and synthetic graphite, and has an average particle size that is $10~\mathrm{nm}$ to $10~\mathrm{\mu m}$ and smaller than the average particle size of the activated carbon.

14. (Cancelled).

15. (Withdrawn) The powder mixture of claim 13, which is composed of 0.1 to 20 parts

by weight of the conductive powder per 100 parts by weight of the activated carbon.

16. (Withdrawn) The powder mixture of claim 13 or 15, wherein the activated carbon has

an average particle size of 0.1 to 100 µm, and wherein the conductive powder adheres to the

periphery of the activated carbon and has an average particle size that is 10 nm to 10 µm and

smaller than the average particle size of the activated carbon.

17. (Withdrawn) The powder mixture of claim 13, wherein the activated carbon has a

packing density of not more than 1.0 g/cm³ and an average particle size of 0.1 to 100 um.

18. (Withdrawn) The powder mixture of claim 13, wherein the activated carbon is

prepared by subjecting a mesophase pitch-based carbon material, a polyacrylonitrile-based

carbon material, a gas phase-grown carbon material, a rayon-based carbon material or a pitch-

based carbon material to alkali activation with an alkali metal compound, then grinding the thus

obtained carbon material

19. (Withdrawn) A polarizable electrode composition prepared by wet mixing the powder

mixture of claim 13 with a binder polymer in a mixing container subjected to rotational and

revolutionary motion.

20. (Withdrawn) The polarizable electrode composition of claim 19, wherein the binder

polymer is an unsaturated polyurethane compound prepared by reacting:

(A) an unsaturated alcohol having at least one (meth)acryloyl group and a hydroxyl group

on the molecule:

(B) a polyol compound of general formula (1) below

3

Docket No.: 0171-0811P

$HO-[(R^1)_h-(Y)_i-(R^2)_i]_a-OH$ (1)

wherein R¹ and R² are each independently a divalent hydrocarbon group of 1 to 10 carbons which may contain an amino, nitro, carbonyl or ether group.

Y is -COO-, -OCOO-, -NR 3 CO- (R^3 being hydrogen or an alkyl group of 1 to 4 carbons), -O- or an arylene group,

the letters h, i and j are each independently 0 or an integer from 1 to 10, and the letter q is an integer which is ≥ 1 ;

- (C) a polyisocyanate compound; and
- (D) an optional chain extender.
- 21. (Withdrawn) The polarizable electrode composition of claim 19, wherein the binder polymer is a polymeric material having an interpenetrating network structure or a semiinterpenetrating network structure.
- 22. (Withdrawn) The polarizable electrode composition of claim 21, wherein the polymeric material having an interpenetrating network structure or a semi-interpenetrating network structure comprises a hydroxyalkyl polysaccharide derivative, a polyvinyl alcohol derivative or a polyglycidol derivative in combination with a crosslinkable functional group-bearing compound, part or all of which compound is the unsaturated polyurethane compound of claim 20.
- 23. (Withdrawn) The polarizable electrode composition of claim 19, wherein the binder polymer is a thermoplastic resin containing units of general formula (2) below

$$\left\{ \begin{array}{c} C - \left(CH_2 \right)_r O \\ O \end{array} \right\}_s$$
(2)

Docket No.: 0171-0811P

Application No. 10/045,084 Docket No.: 0171-0811P

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in which the letter r is 3, 4 or 5, and the letter s is an integer ≥ 5 .

24. (Withdrawn) The polarizable electrode composition of claim 19, wherein the binder

polymer is a fluoropolymer material.

25. (Withdrawn) A polarizable electrode for electrical double-layer capacitors, which

electrode comprises a current collector coated with a polarizable electrode composition

according to claim 19.

26. (Withdrawn) An electrical double-layer capacitor comprising in part the polarizable

electrode of claim 25 and an electrolyte.

27. (Cancelled)

28. (Currently amended) A method for preparing carbonaceous material powder mixture

for electrical double-layer capacitors comprising the steps of:

(1) placing an activated carbon and at least one electrically conductive powder selected

from the group consisting of carbon black, Ketjen black, acetylene black, carbon whiskers,

carbon fibers, natural graphite and synthetic graphite in a mixing container comprising a

planetary mixer, and then

(2) rotating and revolving the container so as to form a dry mixture of said activated

carbon and said conductive powder, wherein the content in said container has an average particle

diameter of 0.1 to 100 μm_{\star}

wherein the activated carbon has an average particle size of 1 to 100 μm, and wherein the

conductive powder adheres to a periphery of the activated carbon and has an average particle

size that is 10 nm to 10 µm and is smaller than the average particle size of the activated carbon.

5